






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
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 Hydrophilic biopolymeric copolyelectrolytes, and biodegradable wound dressings comprising same.

 Hydrophilic biopolymeric copolyelectrolytes comprising (a) a water-soluble linear anionic protein polyelectrolyte component derived from keratin and (b) a water-soluble linear cationic biopolymer polyelectrolyte component derived from at least one biopolymer selected from the group consisting of collagen and a glucosaminoglycan. Hydrogel membranes formed from the copolyelectrolytes are useful as biodegradable dressings for denuded tissue wound sites.

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HYDROPHILIC BIOPOLYMERIC COPOLYELECTROLYTES, AND
BIODEGRADABLE WOUND DRESSINGS COMPRISING SAME

BACKGROUND OF THE INVENTION

This invention relates to novel hydrophilic biopolymeric materials and, more particularly, to the use of such materials in the form of hydrogel membranes as biodegradable dressings for denuded tissue wound sites such as burn wounds and ulcerations.

In the therapeutic procedure for treating an extensively burned patient, devitalized tissue is removed from the burn site, and the debrided areas are covered with a temporary burn wound dressing prior to definitive autografting. The temporary burn wound dressing ideally provides several important therapeutic functions. First of all, it serves as a barrier to prevent loss of water, salts, and proteins from the internal milieu while blocking microbial infection from the environment. Secondly, it serves to improve wound bed base and promote wound closure, thereby facilitating decontamination and regeneration of the wound area. Thirdly, it serves to alleviate pain.

The best wound coverage material is skin itself-- a biologic dressing with a collagenous component rendering it adherent to endogenous fibrin, and a keratinized water proof surface. Biologic dressings in current use include commercially available pigskin heterograft, and living (donor) human or human cadaver homograft material. While human skin has a number of advantages over pigskin for this purpose, there are many problems associated with obtaining, storing, and

using frozen and lyophilized human skin. Skin banks require at least 200 cadavers per year in order to supply existing burn centers. In any case, either of these types of biologic dressings give rise to rejection phenomena which mandate their removal and replacement every 2 to 5 days, or every 2 to 3 weeks if measures are taken to retard the rejection phenomenon. Eventually, however, these dressings must be stripped, often resulting in bleeding and renewed destruction of the graft base.

The body's rejection of foreign biologic dressings has lead to a search for skin substitutes which are either completely synthetic, derived from tissue components, or some combination thereof. Such search has not heretofore met with great success due to the difficulties involved in finding a material exhibiting the proper combination of properties essential for an ideal skin substitute. These properties include rapid, uniform, and strong adherence of underlying tissues; water vapor transport characteristics sufficient to keep the underlying tissues moist without creating pooling; elasticity; durability; intact bacterial barrier characteristics; nonantigenicity and nontoxicity; high permeability to oxygen; capability of being easily applied and removed; easily storable; and relatively inexpensive.

The materials previously proposed as skin substitutes have generally been found to be lacking in one or more of the foregoing properties. The most satisfactory of these materials have consisted of layered composite membranes having an outer layer designed for durability and elasticity, such as silicone or other synthetic polymeric film; and an inner layer designed for maximum adherence, such as collagen, cotton gauze, or Dacron flocking. However, the necessity for these

composite membrane burn wound dressings to be stripped from the wound prior to definitive autografting, poses some difficult design problems. Since these dressings generally depend upon tissue ingrowth into their inner layer for adherence of the wound, complete removal of the dressing is difficult and may require re-debridement before grafting. While it may be possible to overcome this problem by having the inner layer formed of a biodegradable material, such approach has been found to result either in a premature loss of adherence and effectiveness of the dressing before completion of wound healing, or in the formation of considerable scar tissue, unless the biodegradation rate of the inner layer is carefully controlled so as to precisely coincide with the rate of wound healing.

Homogeneous hydrogel membranes formed from the hydrophilic synthetic polymeric material, Hydron, have also been proposed as skin substitutes for use as burn wound dressings. This material combines adherency to dry and moist tissue with the other desirable properties of conformability to a regular contour, elasticity, nonantigenicity, being inert, and providing an effective antimicrobial barrier. Its major drawbacks as a burn wound dressing, however, are its excessive permeability to water and its low degree of durability when washed with water or in the presence of moderate oozing or bleeding. Moreover, its elasticity is too rapidly lost upon drying. For these reasons, it has not proven to be practical for relatively long-term burn wound dressing applications.

In our co-pending European Application No. 83301149.7 (published on 21.09.83 under Publication No. 0089152A) we have described and claimed a hydrophilic biopolymeric copolyelectrolyte which is a significant improvement over previously known biodegradable wound dressings, and which comprises (a) a water-soluble linear anionic protein polyelectrolyte component derived from keratin and (b) a water-soluble linear cationic biopolymer polyelectrolyte component derived from at least one biopolymer selected from a glucosaminoglycan (which is preferably chitosan, more preferably in carboxylate eg acetate form) and collagen. Our prior application, supra, teaches that the weight ratio of the anionic keratin polyelectrolyte component to the cationic biopolymer polyelectrolyte component in the copolyelectrolyte is most suitably within the range of from about 1:1 to about 10:1, and more preferably within the range of from about 2:1 to about 5:1. Examples are presented which illustrate the preparation of such biopolymeric copolyelectrolytes, and of their successful use as wound dressings in clinical trials on animals.

Summary of the Invention

It has now been found, in accordance with the present invention, that hydrophilic biopolymeric copolyelectrolytes which have improved properties as wound dressings are obtainable, following the general teachings of our earlier application, supra, but employing different relative amounts of the anionic and cationic components.

More specifically, the present invention provides a hydrophilic biopolymeric copolyelectrolyte of (a) a water-soluble linear anionic protein polyelectrolyte component derived from keratin, and (b) a water-soluble linear cationic biopolymer polyelectrolyte component derived from at least one biopolymer selected from a glucosaminoglycan and collagen, the weight ratio

of said anionic protein polyelectrolyte component to said cationic biopolymer polyelectrolyte component lying within the range of from 0.001:1 to 16:1 but outside the range of from 1:1 to 10:1.

5 The parent European Application No. 33301149.7, supra, states that when the cationic biopolymer polyelectrolyte component is a mixture of the glucosaminoglycan and collagen, then the weight ratio of the glucosaminoglycan to collagen is preferably within
10 the range of from 0.5:1 to 2:1. In accordance with a feature of the present invention, it has now been found that copolyelectrolytes which have improved properties as wound dressings are obtainable when the weight ratio of the glucosaminoglycan to collagen is within the range
15 of 2:1 to 13:1, and preferably 3:1 to 10:1. Such novel ratios of glucosaminoglycan to collagen can be employed both with the new ratios of anionic protein polyelectrolyte component to cationic biopolymer polyelectrolyte component which are taught herein, as
20 well as with the ratios of these components previously taught in the parent European application. Accordingly, the present invention further provides a hydrophilic biopolymer copolyelectrolyte of (a) a water-soluble linear anionic protein polyelectrolyte component derived
25 from keratin, and (b) a water-soluble linear cationic biopolymer polyelectrolyte component derived from a glucosaminoglycan and collagen, the weight ratio of said glucosaminoglycan to said collagen being within the range from more than 2:1 to 13:1.

Hydrogel membranes comprising the copolyelectrolytes of the present invention exhibit a combination of properties rendering them useful as biodegradable dressings for burn wounds and other denuded tissue wound sites. Such membranes are strongly adherent to underlying tissues, elastic, durable, highly permeable to oxygen, absorbent to wound exudates without losing their durability, have water vapor transport characteristics sufficient to keep the underlying tissues moist without creating pooling, and have intact bacterial barrier characteristics. They may be readily and conveniently applied to the wound site in several alternative modes. By virtue of their biodegradability, they do not require stripping. In the latter stages of wound healing, when moisture through the wound area is much reduced, any remaining copolyelectrolyte material will dry and harden to a protective carapace, which will fall off naturally without leaving any scarring.

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DESCRIPTION OF PREFERRED EMBODIMENTS

The hydrophilic biopolymeric copolyelectrolytes of the present invention are water-insoluble, water-swellaable materials comprising a water-soluble linear anionic protein polyelectrolyte component derived from keratin and a water-soluble linear cationic biopolymer polyelectrolyte component derived from at least one biopolymer selected from the group consisting of a glucosaminoglycan, such as chitosan, and the protein, collagen. Keratin is a protein obtained from sources such as skin, fur, hair, wool, horn, nails, claws, beaks, and scales. It may be readily isolated from its source material and separated into its alpha-

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keratose and gamma-keratose fractions by procedures well known in the art, such as, for example, as described by Widra, *Mycopathologia et Mycologia Applicata*, Volume 30, pages 141-144 (1966) and Rhodes, *et al.*, *Mycopathologia et Mycologia Applicata*, Volume 33, pages 345-348 (1967), incorporated herein by reference. Chitosan is the deacylated form of chitin, which is a glucosaminoglycan obtained as a major constituent of the shells of shrimp, crabs, and lobsters, the cell walls of filamentous fungi, and the exoskeletons of insects. Chitosan is commercially available in the form of fibers, for example, from Sigma-Aldrich Corporation, St. Louis, Missouri. Collagen is a fibrous protein which comprises the major portion of the white fiber in connective tissues of the animal body, particularly in the skin, bones and tendons. It is commercially available in the form of soluble fibers, for example, from Sigma-Aldrich Corporation, St. Louis, Missouri.

The water-soluble derivatives of keratin employed in the anionic polyelectrolyte component of the copolyelectrolytes of the present invention are linear polyelectrolytes in which the keratin moiety is in anionic form. A particularly suitable anionic keratin polyelectrolyte is ammonium keratinate, obtained as the total ammonium hydroxide-soluble fraction of peracetic acid-oxidized human hair, or the alpha-keratose component of this fraction, by the procedures described in the aforementioned Rhodes, *et al.*, article. Due to evidence indicating a higher degree of nonantigenicity, the alpha-keratose form of ammonium keratinate is preferred.

The water-soluble derivatives of the glucosaminoglycan and collagen employed in the cationic biopolymer polyelectrolyte component of the

copolyelectrolytes of the present invention are linear polyelectrolytes in which the biopolymer moiety is in cationic form. Particularly suitable cationic glucosaminoglycan and collagen polyelectrolytes are the
5 carboxylates of these biopolymers, such as their acetates or citrates, obtained by dissolving the biopolymer in an aqueous solution of the corresponding carboxylic acid. Chitosan acetate, collagen acetate, and mixtures thereof, are preferred for use in the
10 cationic biopolymer polyelectrolyte component.

The weight ratio of the anionic keratin polyelectrolyte component to the cationic biopolymer polyelectrolyte component in the copolyelectrolytes of the present invention is to lie within the range of
15 from 0.001:1 to 16:1 and more preferably within the range of from 0.001:1 to 1:1. When the cationic biopolymer polyelectrolyte component is a mixture of the glucosaminoglycan and collagen, the weight ratio of the glucosaminoglycan to collagen is preferably within
20 the range of from 0.5:1 to 13:1, and more preferably within the range of 3:1 to 10:1.

When the water-soluble anionic keratin polyelectrolyte component is contacted in the presence of water with the water-soluble cationic biopolymer
25 polyelectrolyte component (i.e., cationic glucosaminoglycan, cationic collagen, or mixtures thereof), the polyelectrolyte components spontaneously rearrange themselves into a water-insoluble, water-swellaible solid coherent mass. While the precise mechanism of reaction
30 resulting in the formation of these biopolymeric copolyelectrolyte hydrogels is not known with certainty, it is believed that the initial attraction between the two polyelectrolyte components

is due to their opposite net charge, and that closer juxtaposition of the biopolymer molecules then brings into play a variety of steric fitting and chemical bonding and crosslinking mechanisms at multiple sites along the molecules to produce interdigitating co-biopolymers. In any event, their properties and characteristics are totally different from those of their individual components.

In their hydrated form, the biopolymeric copoly-electrolytes of the present invention are stress-durable hydrogels which may be manipulated like a self-annealing paste or putty and thereby formed into membrane sheets, troweled into crevices, or formed in shaped containers or around glass or metal or through perforations. As long as moisture is present, they remain flexible and elastic. As they dry down, they shrink, adhere to flat surfaces, self-anneal and harden to a crystallite form of packed microfibrils. Since they are permeable to and expansible in water, the copolyelectrolytes may be regenerated by the addition of water from their dehydrated form to their hydrogel form, and thereafter reshaped.

Preparation of the biopolymeric copoly-electrolytes of the present invention may suitably be carried out by mixing together aqueous solutions of the anionic keratin polyelectrolyte component and the cationic biopolymer polyelectrolyte component to precipitation end-point and allowing the resulting integral mixture to dry down to a cohesive membrane. This may be done very carefully in vivo. Alternatively, aqueous solutions of the anionic keratin polyelectrolyte component and the cationic biopolymer polyelectrolyte component may be mixed to form a gel which is then applied to the wound and allowed to dry down to form the cohesive membrane. Yet another

alternative is to mix aqueous solutions of the anionic keratin polyelectrolyte component and the cationic biopolymer polyelectrolyte component to form a hydrogel which is allowed to dry on a mold or form; the membrane may then be removed from its forming substrate either by cracking it off in its brittle dehydrated state and forming it into a powder, or by teasing and floating it off in water as a flexible hydrogel membrane. The optimal weight ratio of the anionic keratin polyelectrolyte component to the cationic biopolymer polyelectrolyte component in copolyelectrolytes made by these mixing methods is about 0.2:1 to about 7:1; a range of from about 0.4:1 to about 3.75:1 is preferred.

As an alternative preparative procedure, one of the polyelectrolyte components in solid form may be contacted with an aqueous solution of the other polyelectrolyte component. For example, the cationic biopolymeric polyelectrolyte component in solution may be dried down to a crystallite complex which resembles cellophane sheeting, its thickness and strength varying with the amount of solution used to cover a given area before dry-down. Contacting the cationic biopolymeric polyelectrolyte sheeting with an aqueous solution of ammonium keratinate results in the formation of a copolyelectrolyte hydrogel membrane. This may be accomplished, for example, by swabbing or spraying the ammonium keratinate solution onto both sides of a sheet of cationic biopolymeric polyelectrolyte component and allowing the wetted sheet to dry. Alternatively, the ammonium keratinate solution may be swabbed or sprayed onto one side of a first sheet of cationic biopolymeric polyelectrolyte component; the wetted side of this sheet is then overlaid with a second sheet of cationic biopolymeric polyelectrolyte

component which may be the same or different from the first sheet. This process may be repeated to form a multiple "sandwich" membrane. In yet another embodiment of this invention, the ammonium keratinate solution is first sprayed or swabbed directly onto the open wound which is overlaid with a sheet of cationic biopolymeric polyelectrolyte component to form a copolyelectrolyte hydrogel membrane in situ. In each of these alternate embodiments the optimal weight ratio of the anionic keratin polyelectrolyte component to the cationic biopolymeric polyelectrolyte component is about 0.001:1 to about 0.2:1. Expressed in more convenient terms, about 30-500 mg anionic keratin polyelectrolyte is applied per 1000 sq. cm. surface area of the cationic biopolymeric polyelectrolyte component sheeting.

The biopolymeric copolyelectrolytes may be formulated with various additives, such as, for example, plasticizers or softening agents, antibiotic, antifungal or other pharmaceutical agents, cells, enzymes, antibodies, pigments, or the like, to enhance their properties for a particular end use. Such additives may suitably be incorporated into the copolyelectrolytes either subsequent to their formation or along with one or more of their polyelectrolyte components during their formation. When used in wound dressing applications, for example, the biopolymeric copolyelectrolytes preferably are mixed with a non-toxic plasticizer or softener, such as glycerol, in an amount sufficient to enhance the flexibility and/or adhesion of the dressing. When employing chitosan acetate sheeting as one of the polyelectrolyte components in formulating the biopolymeric copolyelectrolytes, the plasticizer or softener is advantageously incorporated into the chitosan acetate

sheeting, for example, in a weight ratio of plasticizer or softener to chitosan acetate within the range of from about 0.5:1 to about 12:1.

The hydrogel membranes in accordance with the present invention may be formed in a wide range of thicknesses, the optimum thickness varying with the desired end use. Membranes thicknesses of at least about 1 mil will have sufficient strength and durability for most application. For membranes used as wound dressings, a thickness ranging from about 1 to about 7 mils has been found to be particularly suitable. The membrane thickness may suitably be controlled in various ways. For example, varying the concentrations of the polyelectrolyte components in the stock solutions employed in the formation of the copolyelectrolytes will result in corresponding variations in the resulting membrane thickness. Alternatively, separately formed hydrogel layers may be laminated together, for example, with the aid of an intermediate coating of a suitable softener or plasticizer (e.g., a glycerol-water mixture), thereby forming a composite hydrogel membrane. Such composite membranes may be fabricated with their separate hydrogel layers having either the same or different composition. For example, the cationic biopolymer polyelectrolyte component of the copolyelectrolyte may be cationic chitosan in one layer and cationic collagen in another layer.

The combination of properties exhibited by the biopolymeric copolyelectrolyte hydrogel membranes of the present invention render them particularly suitable for use as dressings for burn wounds and other denuded tissue wound sites. Such membranes are rapidly, uniformly, and strongly adherent to underlying tissues by virtue of their shrinkdown from the

fully hydrated state and/or by virtue of their collagen content and resulting linkage to fibrin in the wound bed. They are durable to physical stress and may be thickened as needed to enhance their durability. They have a high degree of absorbancy for serious or bloody exudate, and remain flexible and elastic so long as moisture is present. Their water vapor transport characteristics are such as to allow pervaporation of water at a rate sufficiently high so as to prevent fluid pooling beneath the dressing, and yet sufficiently low so as to maintain the requisite moisture at the wound surface for wound healing cell migration to occur and the requisite moisture within the membrane for maintenance of flexibility and elasticity. The hydrogel membranes are highly permeable to oxygen, allowing air to get into the wound while stopping bacteria. Their microbial barrier function can be further improved by incorporating antimicrobial agents into the dressing, for example, by inclusion in the hydrogel during its formation, by inclusion between hydrogel layers, or by direct spraying or smearing onto the wound dressing as clinical conditions demand.

The materials used in making the copolyelectrolytes and their hydrogel membranes are non-antigenic and non-toxic, and are readily available. Furthermore, the copolyelectrolytes, either in dehydrated or hydrated form, are easily storable at room temperature in polyethylene bags, aluminum foil packs, or plastic dishes after autoclaving, gas, alcohol, or radiation sterilization. Sterilization by ultraviolet radiation or ethylene oxide gas is preferred.

The copolyelectrolyte hydrogel membrane wound dressings may be readily and conveniently applied to the wound site in several different application modes.

For example, the hydrogel may be preformed and applied to the wound site either as a preformed membrane or as a self-annealing paste. Alternatively, the copolyelectrolyte may be preformed and applied to the wound site in dehydrated form, either as a dried membrane or as a powder, and thereafter allowed to hydrate to an adherent conforming hydrogel membrane in situ on the wound site. A further alternative application mode is to individually apply the anionic keratin polyelectrolyte component and the cationic biopolymer polyelectrolyte component to the wound site so as to effect in situ formation on the wound site of the copolyelectrolyte as a hydrogel membrane. In this latter mode of application, both of the polyelectrolyte components may be applied as aqueous solutions, or one may be applied as an aqueous solution and the other in solid form. For example, the wound site may be first sprayed or swabbed with an aqueous solution of ammonium keratinate, and thereafter overlaid with dried chitosan acetate sheeting, preferably including a flexibility-and/or adhesion-enhancing amount of a non-toxic plasticizer or softener, such as glycerol.

The copolyelectrolyte hydrogel membrane wound coverings can remain in place over substantially the entire wound healing period, during which time fluid exudate from the wound is absorbed, and white blood cells and macrophages infiltrate the hydrogel and ultimately dry at the upper air interface to form a protective scab, while entering fibroblasts elaborating collagen fibers bind to the lower moist surface of the membrane. Epidermal cells move in centripetally from the edge of the wound through these collagen fibers to grow over and close the wound in the moist space between the fibroblasts and the white cell-infiltrated membrane. The moist membrane becomes

biodegraded by the skin cells, white cells and macrophages. In the latter stages of wound healing, when moisture through the wound area is much reduced, the remaining membrane will dry and harden to a protective carapace or scab, which will fall off naturally without leaving any scarring. The dried scab may, if desired, be softened and removed by application of a glycerine-water mixture.

While the hydrophilic biopolymeric copolyelectrolytes of the present invention have been described primarily with reference to their utility as hydrogel membrane wound dressings, it will be understood that these novel materials have a wide variety of other potentially important applications. For example, they may be used in conjunction with implantable prosthetic devices, and as scar tissue coverings, sutures, tapes, sustained release drug carriers, and tube linings in bypass surgery. They may be covered with an overgrowth of a layer or layers of cultured epithelial cells to form a synthetic skin wound dressing or graft, or for use as a lining in prosthetic implants and devices.

The invention is further illustrated by way of the following examples.

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EXAMPLE 1

Stock solutions for use in preparing hydrophilic biopolymeric copolyelectrolytes in accordance with the present invention were prepared in the following manner.

Chitosan acetate solution was prepared by continuously stirring 500 mg of practical grade shrimp chitosan fibers (Sigma-Aldrich Corporation, St. Louis,

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Missouri) into 100 ml of cold 0.25% (v/v) acetic acid. The solution was cleared of gross particulate matter by filtration through a 12 - layer gauze pad on a Buchner funnel.

- 5 Collagen acetate solution was prepared by stirring 200 mg of bovine collagen fibers (acid-soluble Type III, Sigma-Aldrich Corporation, St. Louis, Missouri) in 100 ml of cold 0.25% (v/v) acetic acid.

- 10 Alpha-keratose ammonium keratinate solution was prepared as follows. 12 grams of clean, dry, blond human hair, previously degreased and washed, were placed in a 1 liter Erlenmeyer flask containing 320 ml of water. 80 ml of concentrated peracetic acid was added, and the flask stoppered. The contents of the
15 flask were swirled and then placed in a refrigerator for 24 hours with occasional swirling. The bleached, easily stretched and torn ("retted") hair was then freed of the peracetic acid by decantation and thorough washing with separate water rinses. The washed
20 retted hair was then covered with 800 ml of 3 N ammonium hydroxide, and stirred in the cold for 24 hours to solubilize hair keratins. The total soluble protein (TP) fraction was then cleared of solids by centrifugation, discarding undissolved protein and
25 non-protein residual debris. The TP fraction was further clarified through Whatman No. 1 paper, dialyzed against water until the dialyzate wash gave only a faint positive for ammonia with Nessler's reagent, and then Seitz microfiltered. Alpha-keratose was
30 precipitated from the TP fraction by incremental addition of 0.1 N hydrochloric acid while stirring. The precipitate was collected by centrifugation, and the supernatant gamma-keratose, antigenic for rabbits, was discarded. The alpha-keratose precipitate was
35 washed in water, re-centrifuged, and then solubilized

in 0.1 N ammonium hydroxide. A second cycle of precipitation, washing, and solubilization was run on the alpha-keratose before final dialysis against water, microfiltration, and storage in a sterile container.

- 5 The resulting solution contained approximately 7.5 mg of alpha-keratose ammonium keratinate per ml.

EXAMPLE 2

The chitosan acetate and alpha-keratose ammonium keratinate stock solutions prepared in Example 1, were
10 employed in the preparation of a chitosan keratinate copolyelectrolyte. 50 ml of chitosan acetate solution (containing 250 mg of chitosan acetate) and 15 ml of the ammonium keratinate solution (containing 250 mg of
15 chitosan acetate) and 15 ml of the ammonium keratinate solution (containing 112.5 mg of alpha-keratose ammonium keratinate), were mixed together in a plastic Petri dish to precipitation end-point to produce an opaque whitish, sticky, flocculent precipitate. Upon evaporation of the supernatant, the precipitate dried down to form a hard
20 brittle translucent sheet, which was pried or cracked from the bottom of the dish. Upon adding water to the dried material, a self-annealed, tough, stretchable, cuttable, cohesive chitosan keratinate copolyelectrolyte hydrogel membrane was formed.

EXAMPLE 3

25 All three of the stock solutions prepared in Example 1 were employed in the preparation of a chitosan-collagen keratinate copolyelectrolyte. 30 ml of the chitosan acetate solution (containing 150 mg of chitosan
30 acetate), 10 ml of the collagen acetate solution (containing 20 mg of collagen acetate), and 10 ml of the ammonium keratinate solution (containing 75 mg of alpha-keratose ammonium keratinate), were mixed together in a plastic Petri dish to precipitation end-point. Upon
35 evaporation of the supernatant, the precipitate dried down to a crystalline-like sheet. Upon adding water to

the dried down precipitate, a self-annealed, flexible, cohesive chitosan-collagen keratinate copolyelectrolyte hydrogel membrane was formed, which was teased and floated from the bottom of the dish.

- 5 Examples 4-7 illustrate the preparation of chitosan keratinate copolyelectrolyte hydrogel membranes containing a variety of antibiotics.

EXAMPLE 4

- Chitosan acetate sheeting was prepared by mixing
10 2 ml glacial acetic acid, 2 ml glycerol, 40,000 mg gentamycin sulfate (Garamycin, Schering Corp., Kenilworth, NJ), 796 ml water, and 4 g practical grade shrimp chitosan fibers (Sigma-Aldrich Corporation). The mixture was stirred until a solution was formed. The
15 solution was filtered through five layers of cheese cloth and allowed to dry down to form a cohesive solid sheet of chitosan acetate.

- The 9" x 12" x $\frac{1}{2}$ " dried chitosan acetate sheeting was sprayed on both sides with a total of about
20 10 ml of a 5 mg/ml alpha-keratose ammonium keratinate solution prepared in a manner analogous to Example 1, to form a chitosan keratinate copolyelectrolyte hydrogel membrane containing the representative antibiotic gentamycin sulfate.

EXAMPLE 5

- Chitosan acetate sheeting was prepared by mixing
2 ml glacial acetic acid, 2 ml glycerol, 796 ml water, and 4 g practical grade shrimp chitosan fibers (Sigma-Aldrich Corporation). The mixture was stirred to
30 solution and filtered as in Example 4.

- 70 ml of this chitosan acetate solution was poured into a standard plastic Petri dish (3.5" diameter) and 300 mg demeclocycline powder (Declomycin, Lederle Laboratories Div. American Cyanamid Corp., Wayne, NJ) was
35 added with stirring. The mixture was allowed to dry down to a membrane which was then lifted out and sprayed

on both sides with a total of 3 ml of a 5 mg/ml alpha-keratose ammonium keratinate solution prepared in a manner analogous to that of Example 1, to form a chitosan keratinate copolyelectrolyte hydrogel membrane containing the representative antibiotic demeclocycline.

EXAMPLE 6

Chitosan acetate sheeting was prepared by mixing 2 ml glacial acetic acid, 2 ml glycerol, 796 ml water, and 4 g practical grade shrimp chitosan fibers (Sigma-Aldrich Corporation). The mixture was stirred to solution and filtered as in Example 4.

58 ml of this chitosan acetate solution was poured into a standard plastic Petri dish and 1 ml glycerol and 320 mg cefoxitin powder (Mafoxin, Merck Sharp & Dohme, Rahway, NJ) were added with stirring. To this mixture was added 10 ml of a 5 mg/ml alpha-keratose ammonium keratinate solution prepared in a manner analogous to that of Example 1 and containing 125 mg of dissolved tetracycline (Sumycin, Squibb & Sons, Inc., Princeton, NJ). The combined mixture was allowed to dry down to a yellow-brown chitosan keratinate copolyelectrolyte hydrogel membrane containing the representative antibiotics cefoxitin and tetracycline.

EXAMPLE 7

Chitosan acetate sheeting was prepared by mixing 2 ml glacial acetic acid, 2 ml glycerol, 20,000 g gentamycin sulfate (Valley Biologicals Inc., State College, PA), 796 ml water, and 4 g practical grade shrimp chitosan fibers (Sigma-Aldrich Corporation). The mixture was stirred to solution, filtered, and dried as in Example 5.

The 9" x 12" x 1/4" dried chitosan acetate sheeting was sprayed on both sides with a total of 10 ml of a 5 mg/ml alpha-keratose ammonium keratinate solution prepared in a manner analogous to Example 1 and containing 191 mg dissolved carbenicillin powder

(Geocillin, Roerig Div. Pfizer Pharmaceuticals, New York, NY), to form a chitosan keratinate copolyelectrolyte hydrogel membrane containing the representative antibiotics gentamycin sulfate and
5 carbenicillin.

EXAMPLE 8

Patient A, a 54-year-old white female with multiple left leg ulcers diagnosed as pyoderma gangrenosum secondary to Crohn's disease was seen on
10 Dermatology Service, University of Illinois Hospital, after prior unsuccessful treatment. Two of the ulcer lesions were pre-tibial (a shallow lesion 3" in diameter and an elongated deep crater about 3" x 4" long); a
15 third lesion was a deep finger-like ulcer projecting about 1 1/2" into the back of the leg. The patient had been hospitalized for 14 weeks and was receiving high dose prednisone (200 mg daily) for her Crohn's disease and 1% silver nitrate soaks on the ulcers as an
20 antiseptic and to encourage crusting. Silver nitrate was not effective in containing the _____

lesions, nor was a thick zinc oxide paste later applied as an occlusive dressing over the large lesion effective. The patient described accompanying pain as "hot ice picks being driven into the leg".

5 Only the upper shallow pre-tibial lesion, most resembling the experimental surgical wounds produced in rabbits, was treated at first. The lesion was saturated with a sterile 5 mg/ml solution of alpha-keratose ammonium keratinate prepared in a manner
10 analogous to that of Example 1; thereafter a chitosan keratinate copolyelectrolyte hydrogel membrane containing gentamycin sulfate prepared according to Example 4 was laid on the ulcer. This was dressed with a Telfa pad (Kendall Hospital Products Div.,
15 Chicago, IL) and gauze. Obvious success in treatment of the upper shallow ulcer led to gentamycin sulfate chitosan keratinate hydrogel membrane application to the other ulcers with comparable results. Treatment of the deep finger-like ulcer and the large cratered
20 pre-tibial ulcer containing exposed tendon and a small area of exposed bone was then begun. The ulcers were saturated with a sterile 5 mg/ml alpha-keratose solution as above, then overlaid with a UV-sterilized glycerinated chitosan keratinate copolyelectrolyte
25 hydrogel membrane containing gentamycin, prepared according to Example 4. The patient noted absence of tenseness and pain within minutes after the membrane was in place. From this day onward, pre-spraying of lesions with alpha-keratose was discontinued. In-
30 stead, the dried-down 9" x 12" chitosan acetate sheet was simply peeled from its container form, suspended in air by a set of clips, and sprayed on both sides with a total of 10 ml alpha-keratose (5-7.5 mg/ml) solution prepared according to Example 1, allowed to
35 dry in air, then UV-sterilized under polyethylene

(Saran Wrap) sheeting. After about 7 weeks a five-day-old membrane forming a firm carapace on the large lesion had split and lifted off of the healing site which was covered with collagenous fibrotic material.

5 Exudate and drainage into the gauze overwrap was minimal. The patient was ambulatory, free of pain, and discharged from the hospital two weeks later. Follow-up treatment was on an out-patient basis at progressively longer intervals, where soft degraded or

10 hardened non-adherent membrane was simply washed or cut away, the reduced lesion was washed in 3% peroxide, and a new gentamycin sulfate chitosan keratinate copolyelectrolyte hydrogel membrane prepared as above was applied and dressed as usual.

15 Membrane degradation over the collagen-filled granulation base continued for two more months. "Pearling" at the edge of the lesion indicated new epithelial growth. Continued reduction of the lesion by epithelial ingrowth was apparent through the next 6

20 months to an uneventful recovery with complete re-epithelialization.

The successful treatment of these ulcers demonstrates that the membranes of the present invention

1) accelerate healing, 2) cause epidermal closure

25 over a large area, 3) promote healing over a difficult area (i.e., bone), 4) work against a corticosteroid gradient (e.g. high doses of prednisone), and 5) alleviate pain.

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Example 9

Patient B, a 26 year old black female with progressive systemic scleroderma was seen and treated for painful ankle ulcers on Medical Service,

35 University of Illinois Hospital. She had been

unsuccessfully treated during the previous two weeks of hospitalization with whirlpool baths, nitropaste, betadine dressings, and Oxacillin, an oral antibiotic active against the bacteria Staphylococcus aureus found in her lesions. The wounds were washed with 3% peroxide, and glycerinated glucosaminoglycan keratinate membranes (prepared according to Example 4 but omitting the antibiotic) were applied to the lesions on each leg with the usual Telfa and gauze overwraps. She was discharged, pain-free, two days later and thereafter continued her treatment as an out-patient in Special Medicine (Rheumatology Service). The initial application of plain membranes did not "take" due to underlying mixed bacteria consisting of Staphylococcus aureus and Group 3 beta-Streptococci. With application of a chitosan keratinate gentamycin-containing copolyelectrolyte hydrogel membrane prepared as in Example 4, the left ankle lesion incorporated the membrane which was transformed to a scab. It subsequently healed uneventfully. The right ankle lesion remained exudative and now included Enterococci on culture of exudate. A formulation of the chitosan keratinate copolyelectrolyte hydrogel membrane containing 300 mg of Declomycin, to which the mixed flora were susceptible (prepared as in Example 5) was applied to the wound. It formed an adherent hard scab and remained in place for 8 weeks before cracking and friability allowed bacterial re-entry. The patient was seen weekly on an out-patient basis for several months, whilst wearing a cefoxitin tetracycline-containing glucosaminoglycan keratinate polyelectrolyte hydrogel membrane prepared according to Example 6 over the lesion. Over this period, pain was absent. Several months later, however, the patient was re-admitted into hospital for definitive work on the bacteria complicating the healing process. Pseudomonas

aeruginosa and Group B beta-streptococci were found in the lesion. The above organisms, singly and in mixture, proved to be moderately susceptible to cefoperazone sodium by disc test. Accordingly, a membrane containing cefoperazone (CefobidTM, Roerig, New York City) was applied. On the second and third days following this membrane application pus, which proved to be sterile, was expressed from a central slit made in the membrane and the lesion under the membrane was irrigated with 10 ml of an aqueous gentamycin solution (1,000 mcg/ml) before rewrapping in TelfaTM and KerlixTM. The patient was discharged on the following day with a fresh cefoperazone membrane and dressing in place. This was replaced twice at five daily intervals with new gentamycin/cefoperazone membranes. All cultures following the initial application of a cefoperazone membrane were negative, within two weeks no exudate was apparent, the edges of the lesion became sealed, and central darkening under the membrane presaged scab formation over the lesion.

The cefoperazone membranes used in this clinical test were prepared as follows. 65 ml of standard glycan solution was allowed to dry down in a plastic Petri dish. (Cefoperazone solution precipitates and forms its own membrane in glycan solution.) 1 ml of water-reconstituted CefobidTM solution containing 250,000 mcg of cefoperazone sodium was added to 5 ml of alpha-keratose (5 mg/ml) solution. This mixture was poured over the dry glycan sheet and evenly distributed. The glycan swelled and developed a transient cloudiness as it reacted with the cefoperazone/keratose mixture. After dry-down, the clear membrane was harvested and U.V. sterilized under Saran WrapTM before use.

The following procedure was used to prepare the gentamycin/cefoperazone membranes. 1 ml containing 10,000 mcg of gentamycin sulfate (Valley Biologics, Inc.,

State College, PA) was added to 65 ml of standard glycan solution in a Petri dish and allowed to dry down. 1 ml containing 250,000 mcg of cefoperazone sodium (CefobidTM) was mixed with 5 ml of alpha-keratose (5 mg/ml) and 5 evenly distributed over the dried glycan sheet. This was allowed to dry down in turn. The gentamycin/cefoperazone-containing membrane was then harvested and U.V. sterilized under Saran WrapTM before use.

Example 10

Patient C, a 72-year-old white male with a history of rheumatic fever, painful arthritis, a heart valve prosthesis, and an infra-inguinal vessel grafted to treat peripheral vascular insufficiency in his legs had been hospitalized at Lutheran General Hospital, Park Ridge, Illinois, when first seen for leg ulcers. The large pre-tibial ulcer (ca. 3" x 6") on his left leg was grossly contaminated with mixed Pseudomonas aeruginosa and Staphylococcus aureus, showed a gangrenous tendon, and was developing a progressive cellulitis traveling toward the knee. A surgical conference group recommended amputation below the knee which was refused by the patient.

The patient was given a single course of amikacin/prostaphcillin therapy by intravenous drip. No surgical intervention took place. Simultaneously, therapy using 3% peroxide washes followed by application of chitosan keratinate/gentamycin copolyelectrolyte hydrogel membranes prepared according to Example 4 and the usual Telfa and gauze overwraps was initiated. Exudate-wet outer gauze wrapping was changed 2 or 3 times daily. Telfa pads and the chitosan keratinate/gentamycin copolyelectrolyte hydrogel membranes were replaced every 4 or 5 days following a peroxide wash with gauze pad swabbing to remove detritus. Only non-adherent membrane was removed or cut away. The patient was moved to a nursing facility within two weeks where glucosaminoglycan keratinate gentamycin copolyelectrolyte hydrogel membrane therapy was continued. The patient noted absence of pain with the membrane in place. Within 3 weeks the advent of fresh granulation tissue and sealing of the wound edges was apparent. However,

continuing destruction of the devitalized tendon was apparent too. The tendon was soft and freely movable while the attending physician expressed pus for bacteriological studies. It was subsequently lifted and painlessly cut out. The Pseudomonas organism proved to be very sensitive to carbenicillin, and so two weeks after surgery the tendon-free lesion was cleaned with peroxide and swabbing, then covered with a gentamycin/carbenicillin-containing chitosan keratinate copolyelectrolyte hydrogel membrane prepared according to Example 7. Carbenicillin-containing solutions polymerize directly and non-uniformly with the glycan solution. Membranes containing carbenicillin must therefor be formulated by spraying the chitosan acetate sheet (in this instance containing gentamycin sulfate) with alpha-keratose solution containing dissolved carbenicillin powder.

Collagen deposition, development of new granulation tissue, and continued ingrowth of peripheral epithelial tissue is taking place in spite of continued exudation from the center of the lesion. Two smaller ulcers, one on the underside of the left leg (1" x 2") and one on the inner aspect of the right ankle (1" diameter) were successfully closed with gentamycin-containing glucosaminoglycan keratinate copolyelectrolyte hydrogel membranes prepared according to Example 4.

Example 11

Polytetrafluoroethylene (Teflon, E.I. DuPont deNemours and Co., Wilmington, Delaware) tubing is used in surgery as a prosthetic device to replace blood vessels. Narrow-bore tubing is extremely prone

to thrombus formation. Hence there is much interest and value in obtaining a non-thrombogenic lining for such tubing or better yet, a lining which will support confluent growth of endothelial cells, the natural lining of blood vessels.

Such a lining was prepared by the following procedure: A section of polytetrafluoroethylene (PTFE) tubing (Gore-tex, W.L. Gore and Assoc., Flagstaff, Arizona) was soaked in absolute ethanol to destroy its hydrophobic properties. The tubing became translucent; excess fluid was drained but the tubing was not allowed to dry completely.

One end of the tubing was pinched closed and a Pasteur pipette was used to fill the tube with a portion of a solution of 2 ml glacial acetic acid, 2 ml glycerol, 796 ml water, and 4 g practical grade shrimp chitosan fibers (Sigma-Sldrich Corporation). The upper end of the filled tube was then stretched gently in order to insure thorough wetting of all pores in the tube. The filled tube was then laid in a horizontal position and allowed to dry down. The tube as rotated once during drying to insure even dry-down. The efficacy of the chitosan acetate coating procedure was determined with fresh Gram's iodine which wetted and stained (blue-black) the glycan-coated areas only.

With a second Pasteur pipette, the chitosan-acetate-coated lumen of the tube was wetted with a solution of 5 mg/ml alpha-keratose solution prepared in a manner analogous to Example 1; the tubing was then allowed to dry again.

The resulting dried, coated tubing was then used as a substrate for live cell culture. Canine endothelial cells were grown on chitosan keratinate membrane-coated PTFE in Roswell Park Memorial Institute (RPMI) 1640 tissue culture medium (MA

bioproducts, Walkersville, MD). The cells spread more rapidly on such coated PTFE than on uncoated PTFE controls.

5 In order to reduce thrombogenicity in the prosthesis, heparin sodium may be dissolved in the alpha-ketose solution used to wet down and bind the chitosan acetate component to the tube. (Heparin polymerizes directly with the chitosan acetate solution and therefore makes uniform dispersion in
10 chitosan acetate solutions difficult).

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CLAIMS:

1. A hydrophilic biopolymeric copolyelectrolyte of (a) a water-soluble linear anionic protein polyelectrolyte component derived from keratin, and (b) a water-soluble linear cationic biopolymer polyelectrolyte component derived from at least one biopolymer selected from a glucosaminoglycan and collagen, the weight ratio of said anionic protein polyelectrolyte component to said cationic biopolymer polyelectrolyte component lying within the range of from 0.001:1 to 16:1 but outside the range of from 1:1 to 10:1.
2. The copolyelectrolyte of Claim 1, comprising an integral mixture of said anionic protein polyelectrolyte component and said cationic protein polyelectrolyte component.
3. The copolyelectrolyte of Claim 1, comprising one or more layers of said anionic protein polyelectrolyte component, which anionic layers may be the same or different, alternated with one or more layers of cationic protein polyelectrolyte component, which cationic layers may be the same or different.
4. The copolyelectrolyte of any preceding claim, wherein said anionic protein polyelectrolyte component is ammonium keratinate.
5. The copolyelectrolyte of Claim 4, wherein the keratin moiety of said ammonium keratinate is alpha-keratose.
6. The copolyelectrolyte of any preceding claim, wherein said glucosaminoglycan is chitosan.
7. The copolyelectrolyte of any preceding claim, wherein said cationic biopolymer polyelectrolyte component is a biopolymer carboxylate.
8. The copolyelectrolyte of Claim 7, wherein said biopolymer carboxylate is a biopolymer acetate.
9. The copolyelectrolyte of any preceding claim, wherein the weight ratio of said anionic protein

polyelectrolyte component to said cationic biopolymer polyelectrolyte component is within the range of from
5 0.001:1 to 1:1.

10. The copolyelectrolyte of any preceding claim, wherein the biopolymer moiety of said cationic biopolymer polyelectrolyte component is a mixture of chitosan and collagen in a chitosan to collagen weight
5 ratio of from about 0.5:1 to about 13:1.

11. A hydrophilic biopolymeric copolyelectrolyte of (a) a water-soluble linear anionic protein polyelectrolyte component derived from keratin, and (b) a water-soluble linear cationic biopolymer polyelectrolyte
5 component derived from a glucosaminoglycan and collagen, the weight ratio of said glucosaminoglycan to said collagen being within the range from more than 2:1 to 13:1.

12. The copolyelectrolyte of Claim 11, wherein said weight ratio of said glucosaminoglycan to said collagen is within the range from 3:1 to 10:1.

13. A composition comprising the copolyelectrolyte of any preceding claim, and a flexibility- and/or adhesion-enhancing amount of a non-toxic plasticizer or softener.

14. The composition of Claim 13, wherein said plasticizer or softener is glycerol.

15. A composition comprising the copolyelectrolyte of any one of Claims 1-12, and an antibiotic.

16. A hydrogel comprising the copolyelectrolyte of any one of Claims 1-12 in hydrated form.

17. A membrane comprising at least one layer of the hydrogel of Claim 16, said membrane having a thickness of at least about 1 mil.

18. The membrane of Claim 17, wherein both cationic chitosan and cationic collagen are present either in the same hydrogel layer or in two separate hydrogel layers.

19. A biodegradable dressing for burn wounds and other denuded tissue wound sites, comprising the membrane of Claim 17.

20. The dressing of Claim 19, wherein said membrane includes a flexibility- and/or adhesion-enhancing amount of a non-toxic plasticizer or softener.

21. The dressing of Claim 20, wherein said plasticizer or softener is glycerol.

22. The dressing of any one of Claims 19-21, wherein said membrane includes an antibiotic.

23. A biodegradable lining for surgical prostheses, comprising the copolyelectrolyte of any one of Claims 1-18.